

課題番号 : F-14-HK-0060
利用形態 : 機器利用
利用課題名(日本語) :
Program Title (English) : Improvement of Plasmon-Enhanced Water Splitting using Gold Nanostructured SrTiO₃ Single Crystal Photoelectrode with TiO₂ Thin Film Heterojunction
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1. 概要(Summary)

We developed a plasmon-enhanced water splitting system using a gold nanostructured SrTiO₃ single crystal photoelectrode with a TiO₂ thin film heterojunction. Simultaneous evolution of hydrogen and oxygen was observed in separate reaction chambers under visible light irradiation. The substrate composed of single crystal SrTiO₃ with a rutile TiO₂ thin film heterojunction exhibited enhanced water spitting activity compared with the electrode without the TiO₂ thin film because the synergistic effect may prevent the back electron transfer reaction.

2. 実験(Experimental)

• Apparatus

Atomic layer deposition, Scanning electron microscopy, Gas chromatography-mass spectrometer, Gas chromatograph with thermal Conductivity detector, X-ray diffraction, Magnetron sputtering deposition

• Method

A thin gold film was deposited on the front side of the SrTiO₃ by helicon sputtering. The thin film was transformed to discontinuous gold nanoparticles (Au-NPs) after annealing at 800°C for 1 h in a nitrogen atmosphere. As a H₂ evolution co-catalyst, a Pt thin film was subsequently sputtered on the back side of the SrTiO₃ substrate. A TiO₂ thin film was deposited on the back side of a SrTiO₃ substrate by atomic layer deposition prior to the Pt decoration as a comparison.

3. 結果と考察(Results and Discussion)

The water splitting activity was increased with rutile layer decoration. Because the conduction band of rutile is 0.2 eV lower than SrTiO₃. There is a potential gradient between the two sides of the

substrate, under its influence the electrons are transferred from the Au-NPs to the Pt co-catalyst. With the lower conduction band at the H₂ evolution side, the electron transfer is facilitated due to the greater band slope that prevents back electron transfer. For anatase-type TiO₂, whose conduction band energy is almost same as SrTiO₃, the activity decreased. Because of the recombination between electrons and holes are aggravated.

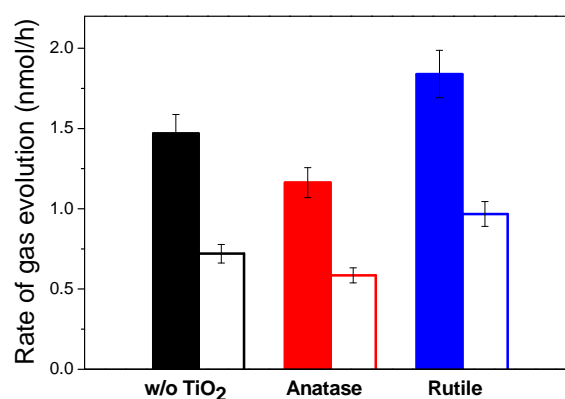


Fig. 1 H₂ and O₂ evolution rates with anatase or rutile layers compared with that without a TiO₂ layer.

4. その他・特記事項(Others)

•Reference

D. Tsukamoto et al., *J. Am. Chem. Soc.* **134**, 6309 (2012).

5. 論文・学会発表(Publication/Presentation)

なし

6. 関連特許(Patent)

出願済み